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EFFECT OF HYDROGEN ON THE MECHANICAL PROPERTIES OF
TITANIUM AND OT4-1 ALLOY

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EFFECT OF HYDROGEN ON THE MECHANICAL PROPERTIES OF
TITANIUM AND OT4-1 ALLOYMoscow Aviation Technology Institute,
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ABSTRACT

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Tests to determine the effect of hydrogen impurity on the mechanical properties of titanium and a low alloy of titanium are described. It turns out that the property most effected by increased hydrogen content is the ductility, which increases the embrittlement and fracture tendency of these metals under dynamic loading. The situation is alleviated, however, by heat treatment at 900°C with quenching in air or water.

It is found that existing Soviet norms for the maximum permissible ductility of the tested metals are not adequate to ensure passable ductility values. Nor is it certain, if the properties are improved by heat treatment, just how stable they will remain with aging. Further investigations are indicated.

AUTHOR

It is assumed at the present time that the ductility of a metal is largely 124* characterized by the tendency of mechanical parts toward brittle fracture under dynamic loading (ref. 1).

The most suitable of Soviet-made titanium alloys for parts to operate under dynamic loading are the low alloy types OT4-1 (1.0-2.5% Al, 0.8-2% Mn) and

*Numbers in the margin indicate pagination in the original foreign text.

about 80 deg/hr, resulting in a final material with an almost equilibrium state.

The principal part of the investigation was conducted with the pure titanium and OT4-1 alloy in the annealed state, except for certain special cases as indicated. /125

TABLE 1

CHEMICAL COMPOSITION OF THE TEST MATERIALS								
Material	Content, %							
	Al	Mn	H ₂	N ₂	C	Fe	Si	O ₂
VT1-1	-	--	0.007	0.0015	0.012	0.06	0.045	0.013
OT4-1	2.4	1.11	0.008	0.030	0.030	0.09	0.07	---

TABLE 2

MECHANICAL PROPERTIES OF THE TEST MATERIALS (FORGED BAR WITH CROSS SECTION 14 x 14 mm)				
Material	Mechanical properties			
	σ_T kg/mm ²	δ , %	ψ , %	a_K kg/mm ²
VT1-1	49.7	26.0	59.3	13.9
OT4-1	76.9	17.2	50.1	6.2

Static tensile tests were performed on the metal with different contents of hydrogen, moving the traverse of the tensile testing machine at standard rates (2 to 4 mm/min), along with ductility tests, using the appropriate equipment and standard samples.

It is evident from the data of table 3 that increasing the hydrogen content to 0.032% in commercial titanium and to 0.015% in OT4-1 alloy elicits essentially no change in the normal mechanical properties of the materials, except in the ductility.

TABLE 3

MECHANICAL PROPERTIES VS. HYDROGEN CONTENT OF THE TEST MATERIALS						
Material	H content, % by wt.	Rockwell hardness	σ_T kg/mm ²	δ , %	ψ , %	α_K kg/mm ²
Scale B						
VT1-1	0.0044	82	49.4	27.0	58.0	15.5
	0.0073	79	45.3	29.2	68.2	6.8
	0.017	85	50.8	29.5	48.6	2.0
	0.032	80.5	45.5	25.8	58.9	0.25
	0.6	81.3	not determined			0.4
Scale C						
OT4-1	0.006	21	08.7	13.8	31.6	3.3
	0.015	20	70.1	14.5	20.0	1.1
	0.6	24	73.4	1.5	2.3	0.3

Note: The mechanical properties were determined from test data for 3 or 4 samples.

However, increasing the hydrogen content to 0.1% sharply lowers the plastic properties of the OT4-1 alloy.

The characteristic most sensitive to changes in the hydrogen content turns out to be the ductility. Increasing the hydrogen content to 0.015-0.017% greatly lowers the ductility of both commercial titanium and OT4-1 alloy (more than sevenfold relative to the vacuum annealed samples). Consequently, the maximum hydrogen content permissible by technical standards (0.015%) does not guarantee satisfactory ductility on the part of commercial titanium or OT4-1 alloy.

An analysis of breaks in impact-tested samples (fig. 1) shows that the observed reduction in ductility with increasing ^{hydrogen} content is accompanied by

enlargement of the fracture facets.¹ In commercial titanium, this is related to an increase in the dimensions of the hydride precipitates, which are essentially internal incipient crevices, along which fracture occurs. With a hydrogen content as high as 0.032%, the hydride phase in commercial titanium is located primarily along the grain boundaries. With a high hydrogen content (0.1%), it emerges in the form of flakes, predominantly inside the grain (fig. 2).

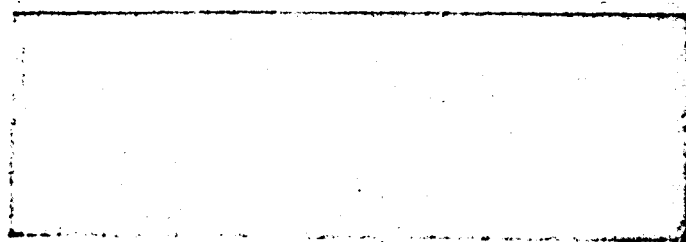


Figure 1. Fractures of Impact-tested Samples of Alloy OT4-1 with a Hydrogen Content of 0.006% (a) and ~ 0.1% (b). X1.5

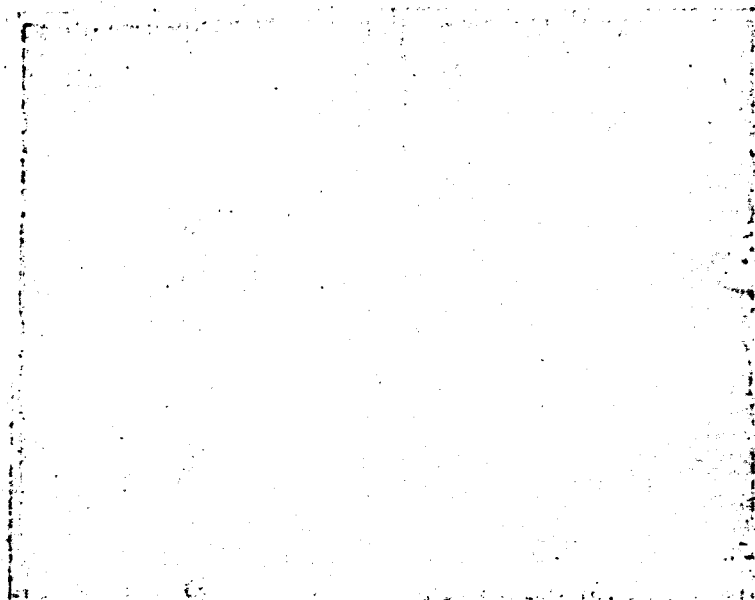


Figure 2. Microstructure of Alloy VT1-1 with a Hydrogen Content of 0.032%, x 400 (a), and ~ 0.1%. x 90 (b).

¹By facets we mean the surfaces along which rupture of the sample occurs.

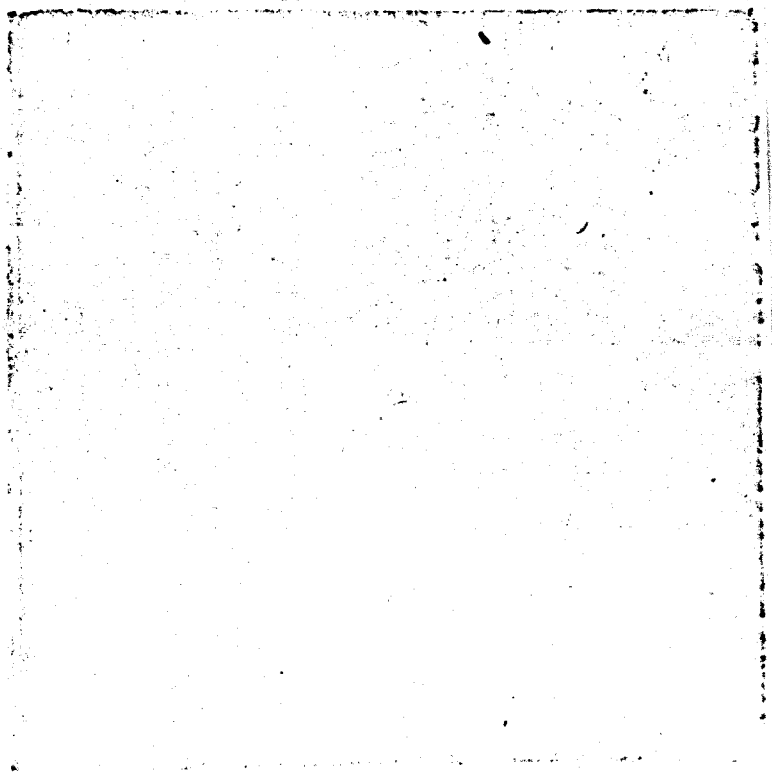


Figure 3. Microstructure of Alloy OT4-1 with Various Hydrogen Contents:

- a) after vacuum anneal (α -phase with streaks of β -phase);
- b) 0.013% (hydride flakes along streaks of β -phase);
- c) 0.1%;
- d) the same in dark field, the arrows indicating the hydride phase at interface between α - and β -phases;
- e) 0.024%, treatment at 900°C with cooling in water;
- f) the same with cooling in air.

a, b, e, f) x 400; c, d) x 600.

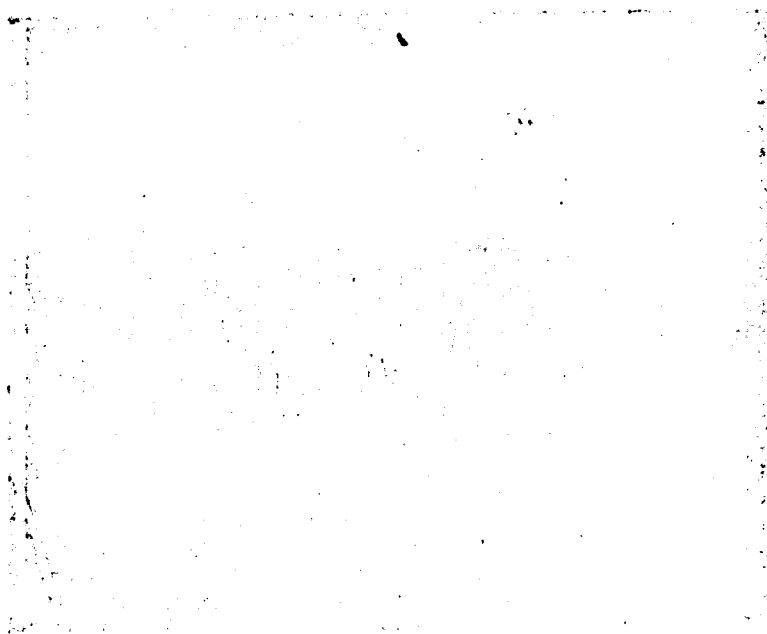


Figure 4. Microstructure at the Fracture Surface of OT4-1 Alloy Sample With a Hydrogen Content of 0.006% (a) and ~ 0.1% (b). x 400.

After annealing, the OT4-1 alloy has a structure consisting of elongated flakes of α -phase, along the boundaries of which occurs a second phase; according to the data of X-ray structural analysis, this phase is a β -solid solution. When examined under the microscope, it has a bright coloring and clearly defined boundaries (fig. 3a), which also show up well in a dark field.

With a higher content of hydrogen, dark-etched hydride precipitates appear at the boundaries between the indicated phases, their number increasing as the hydrogen content is increased (see figs. 3b, 3c). A photograph taken in a dark field discloses that the hydride phase occurs at the boundary between the α - and β -phases (see fig. 3d).

The fracture of ductile samples of OT4-1 alloy subjected to vacuum annealing occurs predominantly through the body of the grain, giving rise to considerable plastic deformation of the metal (fig. 4a). In samples of VT4-1 alloy with a high hydrogen content (~0.1%), fracture proceeds in large measure along the

boundaries of the former grain of the β -phase. In this case, plastic deformation of the flakes of α - and β -phase is not observed (see fig. 4n).

It is known that the hydrogen embrittlement of titanium and its alloys can be abated by heat treatment. It was noted in our investigations that heating of the OT4-1 alloy to 900°C with subsequent cooling in water or air sharply raises the ductility, even with a 0.024% H-content, which far exceeds the permissible limit for its content by technical standards (fig. 5).

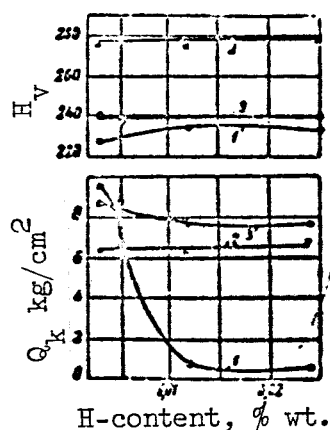


Figure 5. Influence of Hydrogen on the Ductility and Hardness of OT4-1 Alloy After Cooling From 900°C in the Oven (1), in Water (2), and in Air (3).

After cooling in water from a temperature of 900°C, the structure of the metal comprises two phases: a residual α -phase and a martensite α' -phase (see fig. 3e). The presence of the α' -phase in the structure causes greater hardness and somewhat lower values for the ductility relative to the metal cooled in air. The structure of the metal cooled in air comprises flakes of the α -phase 127 (see fig. 3f) with interjacent portions of a second phase, which is clearly β -solid solution. The number of hydride flakes after heat treatment with cooling in air and especially in water is considerably less than in the annealed samples.

The tests that gave the results shown in figure 5 were conducted from 10 to 15 days after heat treatment of the samples. However, in the course of natural or artificial aging of the quenched titanium alloys, it was possible for their plastic attributes to be diminished.

As shown by the data in table 4, OT4-1 alloy with an average hydrogen content of 0.008%, quenched in water from a temperature of 900°C, actually increases in strength after six-day aging and has lower values for the relative elongation and ductility by comparison with the same alloy subjected to vacuum annealing.

TABLE 4

CHANGE IN THE MECHANICAL PROPERTIES OF OT4-1 ALLOY AFTER HEAT TREATMENT						
Heat treatment conditions	Aging conditions	Mechanical Properties				
		σ_T , kg/mm ²	δ , %	ψ , %	H_v	σ_{κ} , kg/mm ²
Vacuum annealing at 900° for 6 hr		68.7	13.8	31.6	229	9.5
900° for 30 min, cooling in water	without aging	73.2	14.3	33.7	269	8.3
	60 hr at 300°	79.0	9.5	25.2	277	7.2
900° for 30 min. cooling in air	without aging	69.3	16.1	40.8	229	10.7
	60 hr at 300°	70.2	14.5	40.1	248	9.1

Artificial aging of OT4-1 alloy from the same melt, first normalized at 900°, did not bring about any reduction in the plasticity characteristics; the 128 ductility in this case was lowered by 1.5 kg/mm², yet stayed at the level of the vacuum annealed metal.

A diminution did not occur in the plasticity or ductility of the metal in large forged pieces of TR4-1 alloy subjected to normalization under the given conditions, nor in the course of subsequent aging in the atmosphere for one year (see table 5).

TABLE 5

MECHANICAL PROPERTIES OF FORGED PIECES OF OT4-1 ALLOY SUBJECTED TO ONE-HOUR NORMALIZATION AT 900°, AFTER AGING UNDER ATMOSPHERIC CONDITIONS FOR ONE YEAR							
Percentage of basic alloying elements		H-content % wt.	Test period	Mechanical properties			
Al	Mn			σ_r kg/mm ²	δ , %	ψ , %	α_k kg/mm ²
1.26	0.94	0.005	before aging	58.7	19.4	40.5	10.3
			after aging	57.0	17.0	38.2	0.8
1.8	1.2	0.007	before aging	59.5	18.4	38.6	9.7
			after aging	58.6	18.2	30.5	9.1

However, these data refer only to metal with a hydrogen content less than 0.010%.

The problem of stability of the plasticity and viscosity during aging of forged or normalized OT4-1 alloy containing 0.015% hydrogen or more will require additional investigations.

CONCLUSIONS

1. Of all the properties investigated, the ductility of VT1-1 and OT4-1 alloys is the most sensitive to changes in the hydrogen content. The brittle fracture tendency of the investigated titanium alloys under dynamic loading, being associated mainly with the ductility, turns out to be smaller, the lower the content of hydrogen in the metal.

2. The upper limit established according to existing technical specifications for the hydrogen content in VT1-1 and OT4-1 alloys (0.015%) does not guarantee the required ductility. Additional investigations using data existing at the plant on the ductility of melts with different hydrogen contents will be required in order to establish the admissible hydrogen content in these alloys.

3. Heating of the OT4-1 alloy to 900° with subsequent cooling in air or water makes it possible to attenuate the detrimental effects of hydrogen on the ductility (its content within the limits up to 0.01% that we investigated). However, additional experiments will be required to study the permanency in natural and artificial aging of the properties acquired by heat treatment.

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